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Violet stimulated luminescence: geo- or thermochronometer?

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Keywords
Violet stimulated luminescence, VSL dating, palaeosol, OSL-thermochronometry, hydrothermal fluids, Golan plateau.

Highlights
- We investigate the potential of VSL to date quartz from early Quaternary palaeosols (1.6 – 0.7 Ma old).
- The VSL signals show good reproducibility, dose response, thermal stability, and dose recovery.
- VSL ages from three palaeosols underestimate the independent K-Ar ages by 50% or more.
- It is possible that the VSL ages are correct, but have been reset by thermal anomalies.
- Further investigation of the natural VSL signal is needed to confirm these underestimating ages.

Abstract
The method of quartz optically stimulated luminescence (OSL) dating is widely used, but generally limited to the past ~0.1 million years (Ma) due to early saturation of the desired signal. Violet stimulated luminescence (VSL) of quartz has previously been shown as a promising alternative, with a dose saturation level ~20 times higher compared to that of OSL, excellent thermal stability on the 10^11 year time scale, and agreement between VSL and OSL ages up to ~0.3 Ma. Here we explore the usability of the VSL signal to date older quartz samples from palaeosols, whose ages are bracketed by K-Ar ages and palaeomagnetic data of the interbedded basalts, emplaced between 1.6 and 0.7 Ma. VSL ages from three palaeosols largely underestimate the independent ages of their overlying basalts. This can be explained either by a low-temperature thermal anomaly resetting the VSL signal in nature, and/or by an insufficient measurement protocol, unable to correctly translate the natural signal into the equivalent laboratory dose.
1. Introduction

Optically stimulated luminescence (OSL) from quartz is a widespread late-Quaternary dating method, used to estimate the time of sediment deposition, and is conventionally measured in the laboratory using blue light stimulation (Wintle and Murray, 2006; Preusser et al., 2009). Although quartz OSL has advantages over other dating methods owing to the ubiquity of the target mineral in geological environments, it is generally limited to ~0.1 Ma (Wintle, 2008). Several attempts have been made to extend this limit, including the use of thermally transferred OSL (Duller and Wintle, 2012 and references therein), slow OSL components (Singarayer et al., 2000), and isothermal thermoluminescence (ITL) (Jain et al., 2005). As initially demonstrated by Jain (2009), stimulation of quartz with violet light after bleaching of the blue-stimulated signal, accesses charge from a deep trap with an extended dose response in the kGy range, by far exceeding that of standard OSL (blue light stimulation). The subsequent work of Ankjærgaard et al. (2013) showed that this violet stimulated luminescence (VSL) signal is associated with a thermoluminescence (TL) trap at ~380 °C with a thermal stability of $10^{11}$ years at 10 °C. The study furthermore reported a coherent VSL chronology agreeing with previous blue OSL ages ($n=9$) on a timescale <0.3 Ma, suggesting VSL as a good candidate for extending luminescence dating to the Ma range. The goal of this paper is to validate quartz VSL dating for the age range up to 1.6 Ma. We present VSL dating results for three quartz-bearing palaeosols bracketed between basaltic flows with K-Ar age constraints between 1.6 – 0.7 Ma (Mor and Steinitz, 1984).

2. Methods and materials

2.1 Instrumentation

VSL measurements were undertaken using a solid-state violet (405 nm) laser diode fitted onto an automated Risø TL/OSL-20 reader by a modification to the reader head, enabling stimulation with both blue (470±30 nm) and violet (405 nm) light. The focusable Nichia 405 nm laser diode had an effective output power of ~35 mW at sample position, giving a uniform illumination area of ~3.5 mm in diameter. To cut off the short wavelength tail, we used a combination of laser bandpass filter 402±15 nm (AHF F49-402) with a 3-mm-thick ITOS GG395 glass filter (ø 18 mm). Luminescence was recorded using a photomultiplier tube (EMI QA 9235), equipped with a 7.5mm Hoya-U340 filter, and a Semrock Brightline 340 nm interference filter (FF01-340/26, ø 45 mm), the latter to minimize violet laser breakthrough into the photomultiplier tube. Samples were irradiated using a $^{90}$Sr/$^{90}$Y source with a dose rate of 0.19±0.01 Gy.s$^{-1}$.

2.2 Study area

Nahal Orvim (33.141N, 35.680E) is situated in the Golan Heights (Israel), adjacent to the Dead Sea transform (DST), see Fig. 1a,b. In this tectonically complex zone the DST splits into a series of branching faults continuing northwards into Lebanon and Syria (Heimann and Ron, 1993). The Golan Heights is a basaltic plateau belonging to the largest volcanic field on the Arabian plate (Harrat Ash Shaam), whose Neogene basalt flows might be petrogenetically related to the tectonic activity on the DST (Weinstein and Garfunkel, 2014). At Nahal Orvim, a basalt sequence is exposed, comprising of several distinct flows, overlying a mid-Eocene limestone at the base of the section. Here we adopt the basalt numbering, NO-1 to 9, of Mor
Figure 1: A+B: Location of the Nahal Orvim site in the Golan Heights (Israel). C: Stratigraphic section at Nahal Orvim, showing K-Ar ages (Mor and Steinitz, 1984; numbers in black), VSL ages (numbers in red), magnetic polarity data (Heimann and Ron, 1993), and VSL sample locations (black circles, this study). Further age constraints arising from combining the K-Ar ages and magnetic polarity data are also indicated by > and < ages. \( \text{§} \)It should be noted that the reported uncertainty on this age might be underestimated as the reproducibility/precision changes of up to \(-60\%\) between repeated sub-sample measurements (Mor and Steinitz, 1984).

and Steinitz (1984) as illustrated in Fig. 1c. The mineralogy of the basalts is alkali-olivine; morphologically, the flows exhibit a massive or columnar structure, and usually feature a vesicular clinker at the base of each flow (Mor and Steinitz, 1984). Some flows are overlain by a red clay-rich and quartz-bearing palaeosol (NOP1-7, Fig. 1c), initially developed on the top of the exposed basalt substrate and subsequently baked and buried under the successive basaltic flow. The high density of the palaeosols is due to dewatering and formation of columnar ‘baked’ structures, often covered with manganese oxides (Singer and Ben-Dor, 1987). Since the basalts are devoid of quartz phenocrysts, the quartz grains embedded in the palaeosols are of a distal aeolian origin (rather than of local, pedogenic, source), and palaeosols with low quartz content are attributed to soil formation during periods of low aeolian input (Singer and Ben-Dor, 1987). Four of the basalt units (NO-1, -4, -6, -9) have been dated by Mor and Steinitz (1984) using K-Ar, bracketing the volcanic eruptions at this site between 1.58 and 0.74 Ma (ages in black, Fig. 1c). These ages are further supported by palaeomagnetic polarity data, sampled at the same site with a higher resolution (Heimann and
Ron, 1993), and attributing three units (NO-3, -4, -5) to the normal polarity of the Jaramillio sub-chron (1.07–0.99 Ma; Singer, 2014, and references within). The combination of K-Ar and palaeomagnetic data allows further refinement of the basalt ages of units NO-2, -3, and NO-5. Newly obtained 40Ar/39Ar ages for the section corroborate the chronology and will be reported elsewhere.

2.3 Sample preparation

Palaeosol samples were collected into light-tight black bags at night under a photographic red lamp, after scraping the exposed outcrops >10 cm inward. The collected materials were then gently crushed in the laboratory, wet-sieved between 75 and 125 µm, and etched with 10% H2O2 and 10% HCl to remove organic matter and carbonates, respectively. The quartz fraction was obtained via magnetic (rather than heavy-liquid) separation in two steps using currents of 0.5 A and 1.4 A to rid of magnetic and iron-coated grains, respectively (Porat, 2006). The resulting quartz fractions were visually checked under the microscope and contained only a few plagioclase feldspar grains, which were further removed by a 20 min 40% HF etching followed by HCl for 40 min.

VSL measurements were carried out on aliquots consisting of quartz grains mounted on silicone-sprayed stainless steel discs, covering the central 2.5 mm diameter. The VSL signals were measured for 500 s; net signals used for analysis were obtained by integrating the first 0-3 s of the optical stimulation, and corrected for ‘background’ taken from last 30 s of the decay curve.

2.4 Dose rate determination

The chemical composition of the sampled palaeosols and basalts was determined at Activation Laboratories Ltd. (Ancaster, Ontario) by inductively coupled plasma (ICP) optical emission spectrometry (OES), and mass spectrometry (MS). Major oxides were determined on a Varian Vista 735-ES ICP-OES; all other elements were measured on a Perkin Elmer Sciex ELAN 6100/9000 ICP-MS. For the radioactive elements used for dose-rate calculation (U, Th and K), we used conservative precision and reproducibility uncertainties of 5% (K2O), 10% (U and Th), see Table 1. Gamma spectrometry measurements confirmed that no significant radioactive disequilibrium was detected in any of the samples. Dose rates to the thin (~20-40 cm) soils were modeled to include a gamma contribution from their adjacent basalt (and in one case, limestone) layers following Aitken (1985). For further details on the dose rate calculations, see the caption to Table 1.

3. VSL signal behaviour

3.1 Protocol and dose response curves

A change in VSL decay curve shape between the natural and the corresponding test dose has been previously reported and discussed in Ankjærgaard et al. (2013). Applying the protocol of Ankjærgaard et al. (2013; listed here in Table 2a) on quartz from Nahal Orvim resulted in significant changes in sensitivity and decay curve shape, motivating us to revert to the protocol of Jain (2009; listed here in Table 2b), and eventually adopt a composite protocol (listed in Table 2c). The main difference of the new protocol compared to Ankjærgaard et al. (2013) is a lower preheat temperature employed for longer time, and separation of the preheat
<table>
<thead>
<tr>
<th>Sample</th>
<th>Age [ka]</th>
<th>U [ppm]</th>
<th>Th [ppm]</th>
<th>K [wt. %]</th>
<th>β [Gy/ka]</th>
<th>γ [Gy/ka]</th>
<th>NO-3</th>
<th>NOP2</th>
<th>NOP3</th>
<th>TOTAL water content [%]</th>
<th>Modelled soil γ dose rate [Gy/ka]</th>
<th>Cosmic dose rate estimated following Prescott and Hutton (1994)</th>
<th>Doubling Alkhen (1985)</th>
<th>Conversion factors from Catten et al. (2011) transforms the measured concentrations to dose rates</th>
<th>Errors on water content are assumed to be 50%. Errors on radioelement concentrations are 5% for K, and 10% for U and Th. Gamma spectrometry measurements on the same samples confirmed no significant radioactive disequilibrium and gave total dose rates well within the 1σ errors of the rates given here.</th>
<th>Dose rates calculated on the basis of measured concentrations and measured concentrations of the basalt and limestone were modelled. The cosmic dose rate is corrected for moisture and loss on ignition (Alkhen, 1985).</th>
</tr>
</thead>
<tbody>
<tr>
<td>Limestone</td>
<td>&gt;5</td>
<td>0.8</td>
<td>0.1</td>
<td>8.3·10⁻³</td>
<td>-</td>
<td>-</td>
<td>0.10 ± 0.01</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td></td>
</tr>
<tr>
<td>NO-1</td>
<td>~4</td>
<td>0.5</td>
<td>1.8</td>
<td>0.62</td>
<td>-</td>
<td>-</td>
<td>0.30 ± 0.01</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td></td>
</tr>
<tr>
<td>NO-2</td>
<td>~8</td>
<td>0.5</td>
<td>2.4</td>
<td>10.8</td>
<td>1.25</td>
<td>39.5</td>
<td>0.018 ± 0.001</td>
<td>1.66 ± 0.08</td>
<td>1.08 ± 0.06</td>
<td>0.71 ± 0.12</td>
<td>1.72 ± 0.23</td>
<td>997.9 ± 36.0</td>
<td>13/13</td>
<td>579 ± 79</td>
<td></td>
<td></td>
</tr>
<tr>
<td>NO-3</td>
<td>~13</td>
<td>1.2</td>
<td>3.5</td>
<td>1.07</td>
<td>-</td>
<td>-</td>
<td>0.56 ± 0.02</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td></td>
</tr>
<tr>
<td>NOP1</td>
<td>~15</td>
<td>0.2</td>
<td>2.2</td>
<td>9.1</td>
<td>1.37</td>
<td>11.2</td>
<td>0.018 ± 0.001</td>
<td>1.69 ± 0.08</td>
<td>1.01 ± 0.06</td>
<td>0.53 ± 0.06</td>
<td>1.88 ± 0.12</td>
<td>406.3 ± 71.4</td>
<td>12/15</td>
<td>216 ± 40</td>
<td></td>
<td></td>
</tr>
<tr>
<td>NOP2</td>
<td>0.2</td>
<td>2.3</td>
<td>10.7</td>
<td>0.79</td>
<td>19.5</td>
<td>0.022 ± 0.001</td>
<td>1.27 ± 0.06</td>
<td>0.95 ± 0.06</td>
<td>0.65 ± 0.08</td>
<td>1.59 ± 0.13</td>
<td>649.4 ± 79.5</td>
<td>13/15</td>
<td>409 ± 60</td>
<td></td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

**Table 1:** Dose rate, equivalent dose, and age information. The three palaeosols investigated are highlighted in bold. The samples are listed in the same order as shown in Fig. 1c. NO-1 denotes a basalt flow and NOP a palaeosol. For samples NO-P: 1-2, the corresponding laboratory numbers are NCL-111295-94-95, respectively.
and the blue OSL wash. The main advantage of the protocol in Table 2c is a reduction in decay curve shape change between the natural and the test dose curves, which can lead to dose underestimations (Ankjærgaard et al., 2013). Further support for the validity of this protocol for the Nahal Orvim samples is obtained via dose recovery tests in Section 3.3.

Fig. 2 shows the VSL dose response for sample NOP2 using protocol 2c. Good recycling within 10% (open symbol), and no significant signal recuperation for the zero dose point are observed. After testing several alternatives, the experimental data are best-fitted with a sum of two saturating exponentials (cf. Ankjærgaard et al., 2013), with two characteristic doses, \( D_{0,1} \ll D_{0,2} \) (Gy). The \( D_{0,2} \) of ~1.6 kGy is higher than that reported by Jain (2009) using protocol 2b, but about half of that reported by Ankjærgaard et al. (2013) using protocol 2a. This difference in \( D_0 \) is the result from the use of different protocols as well as sample-specific behaviour.

![Figure 2: Dose response from sample NOP2 using protocol 2c. To observe the goodness-of-fit at the lower dose range, the same data and fit are repeated on a semi-logarithmic plot (inset). Dashed line corresponds to the best-fit sum of two saturating exponentials, with characteristic doses \( D_{0,1} \sim 100 \text{ Gy} \) and \( D_{0,2} \sim 1600 \text{ Gy} \).](image)

Table 2: VSL protocols used in previous works, and the VSL protocol used in this study.

<table>
<thead>
<tr>
<th>Step</th>
<th>a) SAR-VSL</th>
<th>b) SAR-VSL</th>
<th>c) VSL protocol used in this study</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>Regeneration dose</td>
<td>Regeneration dose</td>
<td>Regeneration dose</td>
</tr>
<tr>
<td>2</td>
<td>Preheat (340 °C, 10 s)</td>
<td>Blue bleach (125 °C, 100 s)</td>
<td>Preheat (300 °C, 100 s)</td>
</tr>
<tr>
<td>3</td>
<td>Blue bleach (280 °C, 100 s)</td>
<td>VSL (50 °C, 500 s)</td>
<td>Blue bleach (125 °C, 100 s)</td>
</tr>
<tr>
<td>4</td>
<td>VSL (125°C, 100 s)</td>
<td>Test dose</td>
<td>VSL (30 °C, 500 s)</td>
</tr>
<tr>
<td>5</td>
<td>Test dose</td>
<td>Cutheat (340 °C, 1 s)</td>
<td>Test dose</td>
</tr>
<tr>
<td>6</td>
<td>Blue bleach (270 °C, 100 s)</td>
<td>Blue bleach (125 °C, 100 s)</td>
<td>Preheat (290 °C, 100 s)</td>
</tr>
<tr>
<td>7</td>
<td>VSL (125°C, 100 s)</td>
<td>VSL (50 °C, 500 s)</td>
<td>Blue bleach (125 °C, 100 s)</td>
</tr>
<tr>
<td>8</td>
<td>Blue bleach (280 °C, 200 s)</td>
<td>Blue bleach (500 °C, 100 s)</td>
<td>VSL (30 °C, 500 s)</td>
</tr>
<tr>
<td>9</td>
<td>VSL (125°C, 100 s)</td>
<td>VSL (50 °C, 500 s)</td>
<td>Violet bleach (380 °C, 200 s)</td>
</tr>
</tbody>
</table>

Repeat steps 1-9, Li Ti
3.2 Thermal stability

The thermal stability of the VSL signal for samples from four different sedimentary environments has been previously reported to have very similar thermal stabilities of \( \sim 10^{11} \) years at 10 °C (Ankjærgaard et al., 2013). In order to compare this to the thermal stability of quartz from Nahal Orvim, and possibly quantify the effects of the new protocol, one aliquot (NOP2) was subjected to thermal holding experiments following a modified protocol 2c, as explained below. Thermally-depleted signals were obtained by giving a dose of 400 Gy, preheating to 300 °C for 100 s, bleaching by blue light at 125 °C for 100 s, and holding at temperature \( T \) [°C] for time \( t \) [s], prior to violet stimulation at 30 °C. A test signal (to correct for potential sensitivity changes) was obtained in a similar fashion, but after a dose 200 Gy, preheat to 290 °C for 100 s, and blue light bleaching at 125 °C for 100 s. Each holding cycle was concluded with a violet bleach at 380 °C for 200 s before repeating the experiment for different \( T \) and \( t \). The resulting isothermal decay curves are shown in Fig. 3; due to a non-first order loss, the effective thermal lifetimes \( \tau_{\text{eff}} \) were obtained via simultaneous fitting of the six isothermal depletion curves with a stretched hyperbolic function (Ankjærgaard et al., 2013, their equation 2). The ‘stretching factor’ \( c \) is a dimensionless parameter common to the six fits, representing the deviation from simple first-order kinetics (where \( c=0 \)). For the fits in Fig. 3, \( c=1.68 \pm 0.07 \), which is similar to the values reported in Ankjærgaard et al. (2013). The inset to Fig. 3 shows the effective lifetime, \( \tau_{\text{eff}} \), plotted against scaled inverse temperature \( 1/(k_B T) \); the slope and the intercept of a regressed straight line corresponds to the VSL thermal trap depth, \( E \) [eV], and the natural logarithm of the escape factor, \( s \) [s\(^{-1}\)]. The values for NOP2 are \( E=1.90 \pm 0.03 \) eV and \( \log_{10}(s)=14.3 \pm 0.3 \) s\(^{-1}\), resulting in a thermal lifetime of \( \sim 7 \cdot 10^{10} \) years at 10 °C, comparable to previously reported values (\( E \sim 1.9 \) eV and \( \log_{10}(s) \sim 14.7 \) s\(^{-1}\), averages from Ankjærgaard et al., 2013), and predicting negligible thermal loss on the timescale of interest for dating (10\(^6\) years).

Figure 3: Isothermal holding of sample NOP2 at temperatures between 240 and 290 °C. Effective isothermal lifetimes \( \tau_{\text{eff}} \) are obtained from simultaneously fitting the decay of each individual temperature with a compressed hyperbola (Eq. 2 in Ankjærgaard et al., 2013). Inset: the linear regression of \( \ln(\tau_{\text{eff}}) \) vs. the inverse of the corresponding holding temperature yields the Arrhenius parameters \( E \) and \( s \) (see text).
3.3 Dose recovery and age estimates

In retrospective dosimetry and in dating, the aim is to estimate the radiation dose absorbed by a sample prior to any laboratory treatment. To test if a measurement protocol is suitable to estimate such doses, a ‘dose recovery’ test is performed; a known laboratory dose is administered to a sample prior to any heat treatment and subsequently recovered using the protocol. If the recovery is successful, the protocol is expected to adequately correct for any sensitivity change during the first SAR cycle (Wintle and Murray, 2006). To test protocol 2c, we performed dose recovery tests on quartz from four palaeosols (NOP1, NOP2, NOP3, NOP5). Samples were bleached by violet light at room temperature for 600 s prior to irradiation with known beta doses between 1 and 3 kGy, which were then treated as unknowns, and recovered using the protocol. The residual dose level following the violet bleach for 600 s was estimated on separate aliquots (NOP1 ~80 Gy; NOP2 ~170 Gy; NOP3 ~40 Gy; NOP5 ~30 Gy) and subtracted from the recovered dose. A dose recovery (one aliquot of sample NOP2; given dose of 3 kGy) is shown in Fig. 4a; the inset illustrates that for this aliquot, no observable change in the decay curve occurs between the given (3 kGy) and the test dose, implying no differential sensitisation between the two measurements (Ankjærgaard et al., 2013). Aliquots showing no growth in the dose response and/or with a recycling ratio deviating with more than 15 % from unity were rejected. Dose recovery experiments are summarized in Fig. 4b; the recovered doses are all normalized by their known given doses, and should be considered successful if the ratio lies within 10 % of unity (Murray & Wintle, 2006). All results fall within 2σ of unity, and for three out of four samples (NOP1, NOP2, and NOP5), the dose recovery ratio is within 10 % of unity. Sample NOP3 overestimates the given dose by 19%, despite absence of obvious signal anomalies. Overall, we interpret the dose recoveries as successful, indicating that the protocol can adequately estimate known doses.

To test the performance of the VSL chronometer on the Ma timescale, we focused on dating the three oldest palaeosols which, unlike the layers above, contain abundant quartz. Equivalent dose estimates were obtained by averaging over the accepted aliquots and are listed with one standard error of the mean (Table 1). The resulting VSL ages all significantly underestimate the independent age control (red numbers in Fig. 1c). NOP1 is dated to $0.22 \pm 0.04$ Ma, in contrast to the overlying basalt age of $1.58 \pm 0.12$ Ma. NOP2 has an age of $0.58 \pm 0.08$ Ma, but is capped by a $>1.07$ Ma basalt, and finally, NOP3 is dated to $0.41 \pm 0.06$ Ma, although it developed before $1.07-0.96$ Ma. Compared to the independent age constraints, the VSL ages underestimate by $>50$ % (NOP2 and NOP3), and $>80$ % (NOP1), despite well-behaved VSL characteristics of signal shape reproducibility, dose response, thermal stability and dose recovery. While only based on three samples, the observed age underestimation seems to decrease with elevation. If these ages are correct, they may reflect a thermal history (discussed in Section 4); however, there also remains the possibility that the ages themselves are erroneous (discussed in Section 5).
4. Thermochronological interpretation of age underestimates

In the above comparison of VSL dates against independent basalt ages, we made an implicit assumption that both chronometers should date the same event of basalt emplacement, corresponding to the simplest geological scenario. However, the two chronological methods (VSL and K-Ar) use signals with very different thermal stabilities: while Ar retention in basaltic glass occurs almost immediately after its solidification at ~500 °C (cf. Caroll, 1991), the accumulation of trapped electrons (as quantified by e.g. OSL, VSL, TL) typically starts hundreds of degrees lower (Guralnik et al., 2013). Thus, a more complex scenario is possible, in which VSL signals are thermally reset by relatively low temperatures, having no effect on the K-Ar dates. Low thermal stabilities of trapped charge may in fact be used for tracing the cooling histories of rocks, referred to as luminescence thermochronometry (Herman et al., 2010; Guralnik et al., 2013).

In thermochronometry, it is recognised that a multitude of cooling pathways may produce exactly the same age, well represented by three extreme scenarios (Wu et al., 2015).
(i), implicitly discussed before, is that of a nearly instantaneous cooling of the baked palaeosol to typical surface temperature of ~30 °C. In this scenario, its VSL age should indeed replicate the K-Ar age of its overlying basalt. However, two additional scenarios exist: (ii) that of a monotonic cooling history, and (iii) of an indefinite storage at an elevated temperature \( T_{app} \). The above three scenarios are the simplest, and have analytical solutions (Guralnik et al., 2013); more complicated scenarios (e.g. a periodic heating) may also produce the observed ages, but require additional constraints from other thermochronometric systems (Reiners and Brandon, 2006). The conversion of a measured trap filling ratio \( n/N \) into an equivalent storage temperature \( T_{app} \) requires knowledge of the natural electron trapping rate \( \dot{D}/D_0 \), and the Arrhenius parameters \( E \) and \( s \); to constrain the closure temperature \( (T_c) \) of a monotonically cooling system, one additionally needs to know the cooling rate (Dodson, 1973; Guralnik et al., 2013).

![Figure 5: Predicted closure temperature as a function of cooling rate for VSL (black), apatite U-Th/He (AHe, grey), and apatite fission track (AFT, grey) thermochronometers. See text for kinetic parameters of the VSL system, and Reiners and Brandon (2006) for thermal stability of AHe and AFT. Notice the similarity between the predicted closure for VSL and AHe systems across a wide range of cooling rates. The thin line segment (in VSL) corresponds to a region where VSL signals would appear saturated, yielding only minimum ages (i.e. allowing to obtain a maximum cooling rate only).](image)

Thermal scenarios (ii) and (iii) are further demonstrated for sample NOP2. The VSL dose response is best described by a sum of two saturating exponentials (Fig. 2). Simultaneous fitting of VSL signals from three aliquots yields a best-fit of \( D_{0,1}=87\pm11 \text{ Gy} \) (28% relative contribution) and \( D_{0,2}=1487\pm115 \text{ Gy} \) (72% contribution). Since the equivalent dose of this sample is 998\(\pm\)36 Gy (see Table 1), we can assume that the lower \( D_{0,1} \) component is thermally stable and in saturation, while the higher \( D_{0,2} \) is only filled to a level of 16\(\pm\)2% of full saturation. The apparent storage temperature can be found using Equation A.10b from Guralnik et al. (2013):

\[
T_{app} = -\frac{E}{k_B} \ln \left[ \frac{N}{n - 1} \right] \frac{\dot{D}}{D_0 s}
\]
Using the filling ratio \( \frac{n}{N} = 0.16 \pm 0.02 \), together with the \( E \) and \( s \) given in Fig. 3, and a \( \dot{D} = 1.72 \pm 0.23 \) Gy/ka (Table 1), the apparent storage temperature can be estimated to \( T_{\text{app}} = 83 \pm 7 \) °C. Considering the alternative scenario of monotonic cooling, VSL closure temperatures \( (T_C) \) are calculated for this sample (following Guralnik et al., 2013) and shown in Fig. 5 for a range of cooling rates, alongside closure temperatures of apatite U-Th/He (AHe) and apatite fission track (AFT) systems (Reiners and Brandon, 2006). The resemblance of the VSL and AHe closure temperatures indicates comparable thermal stabilities of the two methods; for example, for a reference cooling rate of 10 °C/Ma, the closure in both systems occurs at \( T_C \sim 70 \) °C. However, unlike in the AHe method, the VSL chronometer is further subject to signal saturation, which allows to constrain the minimal possible cooling rate (Wu et al., 2015). For the observed trap filling of \( \frac{n}{N} = 0.16 \), and a final cooling temperature of \( \sim 30 \) °C (representing typical daytime temperatures in the Mediterranean climate), the monotonic cooling rate of sample NOP2 can be constrained to \( >4400 \pm 1300 \) °C/Ma, with an associated \( T_C \) of \( >107 \pm 10 \) °C.

5. Discussion

To evaluate the thermochronometric interpretation, we start by noting that the ad-hoc deconvolution of the VSL growth curve into two saturating exponential components is purely empirical, lacking any physical validation. The presence of two such components in the growth curve would imply two charge reservoirs with different capacities yet with an assumed identical thermal stability. Whether such distinct reservoirs exist or not, and whether they correspond to different electron traps, remains to be addressed in the future, and we plan to do so. Our current thermochronometric reconstruction is based on the filling ratio of the non-saturated \( (D_{0.2}) \) component. If we assume that the separation into components is reasonable, we can further compare the obtained results to independent geological constrains.

The reconstructed storage temperature for sample NOP2 (~80 °C) is about ~50 °C higher than typical surface temperatures at the Nahal Orvim site today; it is therefore unlikely, that the NOP2 sample experienced a continuous storage at ~80 °C during the past ~0.5 Ma. The alternative interpretation of a continuous regional bedrock cooling at ~4000 °C/Ma is also unrealistic. Although the Golan Plateau experiences uplift relative to the subsiding Hula basin, cooling rates due to faulting and erosion are expected to be ~3 orders of magnitude slower (Heimann and Steinitz, 1989). To summarize, if we are to accept the thermal resetting of VSL signals at Nahal Orvim, neither the constant storage at elevated temperature, nor the monotonic cooling scenarios apply. On the other hand, these data may be explained by discrete heating events in the past. Such events at an active rift margin are not unlikely: anomalously high water temperatures have been reported from the Shamir artesian drillhole, just 2 km northwest of the Nahal Orvim site (Siebert et al., 2014), as well as 40 km south of the Nahal Orvim site (Gvirtzman et al., 1997; Roded et al., 2013; Shalev et al., 2013).

A similar study undertaken by Ogoh et al. (1993) in a volcanic setting in New Mexico, reported ESR age underestimations of the eruptions ages at Valles Caldera from fission track, U-Th, and \(^{40}\text{Ar}/^{39}\text{Ar} \), arguing that the ESR signal intensity was possibly partially reduced by several later thermal events. This hypothesis was further supported by Lepper and Goff (2007) measuring quartz OSL and polynminerl fine silts IRSL from baked colluvial sediment layers from the same site, obtaining ages in agreement with those of ESR. More recently,
similar age underestimates were observed in a comparable volcanic environment by Tsukamoto et al. (2011) obtaining IRSL ages from plagioclase in bulk basalts measurements. All these studies support the possibility that basaltic sequences in volcanically active areas undergo repeated heating events both due to contact with hot lava, but more importantly due to heating with thermal fluids, which may persist in that region long after the surface basalts are cooled. However, the present VSL data should only be taken as a hint to past thermal events, and not as their proof. Whether or not temperatures as high as 80 °C occurred at the site in the past 1 Ma, and whether the separation into growth components is justifiable, remains to be further explored, e.g. using apatite U-Th/He (Fig. 5) and/or additional luminescence experiments.

Although thermal resetting of the VSL signal could be a cause for the apparent age underestimations, an equally plausible explanation is the potential inability of the VSL measurement protocol (Table 2c) to convert the natural signal into a palaeodose. One of the main assumptions in dating is that the growth of the dose received in nature can be regenerated in the laboratory via an appropriate protocol to measure the dose response curve (Wintle and Murray, 2006). This assumption is rather difficult to test, because it requires a site providing known age samples which are evenly distributed over the dose range of interest and which have small variations in dose rate between samples. Only a handful of studies have so far attempted to evaluate this assumption for quartz OSL (Chapot et al., 2012; Timar-Gabor and Wintle, 2013; Lai and Fan, 2014). Based on blue OSL signals measured in loess stratigraphies, all of the above studies observed that for doses exceeding ~150 Gy, the natural and laboratory-regenerated dose response curves diverge, causing a progressive underestimation of ages at all larger doses. As Ankjærgaard et al. (2013) estimated VSL doses of up to ~200 Gy in agreement with those obtained from OSL, perhaps there is a similar tendency for the natural and regenerated VSL dose response curves to deviate for higher doses, thereby causing the observed underestimations. The ages in Nahal Orvim span between 1.6-0.7 Ma (Fig. 1c) with expected doses ranging between 3-1 kGy; the lack of younger palaeosols capped by basalts at this site limits the characterisation of the natural VSL dose response curve only to high doses. Therefore, the two assumptions that (i) the natural VSL growth curve has the same shape as that measured in the laboratory and (ii) both the natural and regenerated VSL growth curves can be described by a sum of saturating exponentials (as discussed above), can be neither verified nor disproved at this stage. If the natural and regenerated signals with dose are found to deviate, this could easily explain the large dose underestimation, but if they are found to agree, this would be in support of the thermal resetting scenario. For further progress with the VSL method, it is therefore imperative to investigate the natural VSL signal growth curve shape; a test study in a more suitable geological site is currently in progress.

6. Conclusion
In this work we investigated a classic geological section of nine basaltic flows interbedded with palaeosols, with the aim of validating VSL dates from the quartz in the soil layers against independent basalt K-Ar ages. Despite well-behaved VSL characteristics of signal shape reproducibility, dose response, thermal stability, and dose recovery, all three of the dated samples substantially underestimate the soil baking events by ~50% or more. Two possible
interpretations of such a severe age underestimation are (i) the ages are correct, but have been reset by thermal anomalies not affecting the more thermally stable K-Ar system, or (ii) the ages are erroneous, due to the inapplicability of the VSL protocol and/or perhaps sample-specific issues. To decide which of these interpretations is correct, either the apatite U-Th/He and fission track thermochronometers should be further explored, or the natural VSL growth curve shape should be determined. Unfortunately, the latter is not possible at the study site due to lack of soil layers younger than 0.8 Ma, needed to constrain the low-dose VSL response.

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