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Effects of the 2014 major Baltic inflow on methane and nitrous oxide dynamics in the water column of the central Baltic Sea

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Abstract. In late 2014, a large, oxygen-rich salt water inflow entered the Baltic Sea and caused considerable changes in deep water oxygen concentrations. We studied the effects of the inflow on the concentration patterns of two greenhouse gases, methane and nitrous oxide, during the following year (2015) in the water column of the Gotland Basin. In the eastern basin, methane which had previously accumulated in the deep waters was largely removed during the year. Here, volume-weighted mean concentration below 70 m decreased from 108 nM in March to 16.3 nM over a period of 141 days (0.65 nM d⁻¹), predominantly due to oxidation (up to 79 %) following turbulent mixing with the oxygen-rich inflow. In contrast nitrous oxide, which was previously absent from deep waters, accumulated in deep waters due to enhanced nitrification following the inflow. Volume-weighted mean concentration of nitrous oxide below 70 m increased from 11.8 nM in March to 24.4 nM in 141 days (0.09 nM d⁻¹). A transient extreme accumulation of nitrous oxide (877 nM) was observed in the deep waters of the Eastern Gotland Basin towards the end of 2015, when deep waters turned anoxic again, sedimentary denitrification was induced and methane was reintroduced to the bottom waters. The Western Gotland Basin gas biogeochemistry was not affected by the inflow.

1 Introduction

The Baltic Sea is a shallow, semi-enclosed brackish water body. It receives large fresh water inputs from the rivers along its coast, but also exchanges saline water with the North Sea through the narrow Danish straits, principally via the Darss Sill and the Drogden Sill (Fig. 1). This leads to a semi-permanent stratification between relatively fresh surface waters and denser, more saline deep waters. Although intermediate-depth water masses in the southern areas of the Baltic Sea are ventilated frequently, deep waters of the central Baltic are renewed only during major Baltic inflow (MBI) events, during which large amounts of saline oxygen-rich water enter the Baltic through the Danish straits over a short period of time (Schinke and Matthäus, 1998). These events require a specific sequence of weather conditions, occur exclusively in winter and have been occurring approximately once a decade in the recent past (Gräwe et al., 2015).

Due to the semi-permanent stratification, waters below the halocline of the Baltic Sea are typically anoxic (Carstensen et al., 2014) and contain large inventories of reduced compounds produced by microbial and abiotic reactions in the absence of oxygen (Neumann et al., 1997). Many biogeochemical processes are also active in the hypoxic boundary layers close to the halocline (Yakushev et al., 2007; Dellwig et al., 2010). When oxygen is introduced by MBIs, large quantities of the previously accumulated reduced compounds in the deep waters are subsequently oxidized.
(Reissmann et al., 2009) and new redox fronts develop between new and old water masses (Schmale et al., 2016). As such, MBIs have a strong influence on many biogeochemical processes in the Baltic Sea.

In this study we focus on two gases that are strongly influenced by the spatial distribution of hypoxia and anoxia in the Baltic Sea: methane (CH$_4$) (Schmale et al., 2012) and nitrous oxide (N$_2$O) (Hietanen et al., 2012). Both are important greenhouse gases, which also have effects on atmospheric chemistry (Crutzen, 1974; Cicerone and Oremland, 1988). Ambient oxygen concentrations regulate the microbial processes involved in the production and consumption of CH$_4$ and N$_2$O. CH$_4$ is produced in sediments in vast quantities by a unique group of archaea called methanogens (Balch et al., 1979). Methanogenesis is the lowest energy-yield pathway of the anaerobic decay of organic matter, which typically occurs when all other electron acceptors have been depleted (Thauer, 1998). The primary methanogenesis pathway in marine sediments is CO$_2$ reduction by hydrogenotrophic methanogens, while fermentative acetotrophic methanogenesis is the dominant pathway in freshwater sediments (Whiticar et al., 1986). Typically, most of the produced CH$_4$ is consumed in anaerobic and aerobic processes in both sediments and the water column before it can escape to the atmosphere (Reeburgh, 2007; Knittel and Boetius, 2009). N$_2$O is produced by prokaryotes through both oxidative and reductive pathways: as a side product in nitrification (oxidation of ammonium, NH$_4^+$, to nitrate, NO$_3^-$), and as an intermediate product in denitrification (reduction of NO$_3^-$ to elemental nitrogen gas, N$_2$) under suboxic conditions (Anderson and Levine, 1986; Goreau et al., 1980; Bakker et al., 2014). The main biological pathways of N$_2$O production are highly dependent on the oxygen conditions and the availability of organic matter, nitrite (NO$_2^-$) and nitrate (Ward, 2013; Murray et al., 2015). In seas, nitrification has been considered the primary pathway of N$_2$O production (Freng et al., 2012), but recent studies have suggested that the role of incomplete denitrification in the oceanic oxygen minimum zones might have been previously underestimated (Babbin et al., 2015; Ji et al., 2015). Furthermore, it has been shown that archaeal nitrification dominates oceanic N$_2$O production and that this is much more sensitive to oxygen concentrations than bacterial nitrification (Löschler et al., 2012; Freng et al., 2012). N$_2$O is consumed exclusively under anoxic conditions during denitrification (Goreau et al., 1980; Wrage et al., 2001; Bakker et al., 2014). The highest accumulations of N$_2$O in marine waters are typically associated with spatial and temporal redox gradients, such as the margins of oceanic oxygen minimum zones (Naqvi et al., 2010) and dynamic systems including estuaries (Brase et al., 2017).

In the Baltic Sea, surface water concentrations of both gases typically exceed equilibrium with the sea-level atmosphere, indicating supersaturation and an efflux of CH$_4$ and N$_2$O from surface waters (Bange, 2006; Gülzow et al., 2013). However, the sub-haloine profiles of the two gases differ markedly. Deep waters are typically strongly enriched in CH$_4$ below the halocline during stagnation periods (Bange et al., 2010; Jakobs et al., 2013), while N$_2$O is usually absent (Brett and Rheinheimer, 1991). Increased anthropogenic nutrient loading during the last century has been linked to the enhanced production of both gases in the Baltic Sea (Bange, 2006). In the global CH$_4$ budget, the flux from the oceans to the atmosphere is estimated at approximately 2 % of total CH$_4$ emissions (Judd et al., 2002; Reeburgh, 2007), but estimates as high as 10 % have also been presented (Cicerone and Oremland, 1988). Shallow coastal areas and estuaries are an important component, potentially contributing up to 75 % of the total oceanic CH$_4$ flux (Bange et al., 1994). In the case of N$_2$O, the oceans are a much more important source in the global atmospheric budget, contributing up to 25 % of emissions (Nevison et al., 2003). However, there are large uncertainties in the role of coastal zones and estuaries. Previous studies have reported ranges from 11 to 60 % of the total N$_2$O flux from the marine environment (Bange et al., 1996; Seitzinger and Kroeze, 1998). In a more recent study, Naqvi et al. (2010) report a more conservative estimate of 16–33 %, though concede that large uncertainties still exist.

In late December of 2014 a large MBI occurred, during which 198 km$^3$ of water, containing 4 Gt of salt and 2.04 × 10$^4$ t of oxygen, entered the Baltic through the Darss sill (Fig. 1). The inflow was the third largest MBI observed since 1880 (Gräwe et al., 2015; Mohrholz et al., 2015) and caused large changes in the dissolved oxygen concentrations throughout the southern and central Baltic Sea (Mohrholz et al., 2015). The inflow strongly impacted the vertical distribution of CH$_4$ in the Gotland Basin (Schmale et al., 2016), which is the largest sub-basin of the Baltic Sea. Both advective processes (displacement and dilution of old CH$_4$-rich deep waters by the CH$_4$-poor inflow water) and aerobic oxidation of CH$_4$ (stimulated by mixing at the contact between these water masses) contributed to a decline in [CH$_4$] during 2015.

Here we present a broader investigation of the spatial and temporal evolution of both CH$_4$ and N$_2$O concentrations following the inflow, along a transect of sites in both the Eastern and Western Gotland Basin. We discuss the roles of advection and microbial processing in the observed distributions, and the timescales of change in biogeochemical processes in response to the perturbation caused by the MBI.

2 Materials and methods

Samples were collected on six cruises that took place between March and December 2015 on R/V Aranda and R/V Salme. Sampling covered the whole Gotland Basin, with three stations in the Eastern Gotland Basin (EGB: BY10, BY15 and BY20) and two stations in the western basin
The gas samples were analysed within 5 months from sampling using an Agilent Technologies 7890B gas chromatograph with a flame ionization detector (FID) for CH₄ and an electron capture detector (ECD) for N₂O, a 2.4 m HayeSep Q column with 1/8″ connection, 80/100 mesh range and a 1.0 mL sample loop. Carrier gases were nitrogen and helium at 21 mL min⁻¹ flow rates, for N₂O and CH₄ respectively. The oven temperature was 60°C, and detection temperatures for FID and ECD were 250 and 300°C respectively. The samples were injected using a Gilson GX-271 auto sampler. Raw peak area data were converted to mole fraction (ppm) using linear calibration with standards. For CH₄, a three-point calibration was used (0.46, 5 and 47 ppm), consisting of a standard gas mixture (AGA Gas AB, Lidingö, Sweden) with 5 ppm ± 2% CH₄, which was diluted 1:10 with 5.0 purity N₂ to create the low standard. For the high standard, a 1000 ppm ± 2% CH₄ gas mixture (Air Products PLC, Surry, UK) was diluted 1:20 with 5.0 N₂. For N₂O, a two-point calibration (0.10 and 1.1 ppm) was used, consisting of a standard gas mixture (AGA Gas AB, Lidingö, Sweden) with 1.1 ppm ± 5% N₂O, which was then diluted 1:10 with 5.0 N₂. Standard series were analysed prior to each analysis sequence (length of sequence was 40–120 samples) and fitted linearly for each sequence separately to correct for between- sequence drift. Standards containing 5 ppm CH₄ and 1.1 ppm N₂O were analysed after every 10 samples to monitor within- sequence drift (observed to be negligible).

Total in situ dissolved gas concentrations (Cₗ) in mol L⁻¹ were calculated from measured wet mole fraction values in the headspace gas considering the ideal gas law and Henry’s Law:

\[ C_\text{tot} = \frac{P_{\text{HS}} \times V_{\text{HS}}}{R \times T} + F \times P_{\text{HS}} \times V_{\text{Aq}}, \]

where \( P_{\text{HS}} \) is the partial pressure of the gas in the headspace (atm), \( V_{\text{HS}} \) and \( V_{\text{Aq}} \) the syringe headspace and water volume respectively (L), \( R \) is the gas constant (0.08206 L atm K⁻¹ mol⁻¹), \( T \) the temperature during equilibration in Kelvin (293 K), and \( F \) is a salinity- and temperature-dependent equilibrium solubility coefficient of a given gas in mol L⁻¹ atm⁻¹, as defined in Wiesenburg and Guinasso (1979) for CH₄ and in Weiss and Price (1980) for N₂O (in situ salinity and 293 K were used in the calculations).

The calculation can be simplified by assuming that the total pressure in the syringe during equilibration is 1 atm, from which follows that \( P_{\text{HS}} = P_{\text{atm}} \times X_{\text{HS}} \), where \( P_{\text{atm}} \) is the true atmospheric pressure during sampling in atm. Thus, the total original concentration of dissolved gas in the fluid sample can be calculated by summing the contributions from the headspace (\( C_{\text{HS}} \)) and the dissolved phase (\( C_{\text{Aq}} \)):

\[ C_\text{tot} = \frac{(X_{\text{HS}} \times P_{\text{atm}} \times V_{\text{HS}})}{(R \times T \times V_{\text{Aq}})} + F \times X_{\text{HS}} \times P_{\text{atm}}, \]

\[ C_{\text{Aq}} = \frac{C_\text{tot} - \frac{(X_{\text{HS}} \times P_{\text{atm}} \times V_{\text{HS}})}{(R \times T \times V_{\text{Aq}})} - F \times X_{\text{HS}} \times P_{\text{atm}}}{C_{\text{HS}}}. \]
where $X_{\text{HS}}$ is the mole fraction of the gas measured from the syringe headspace in ppmv.

Pre-evacuated Exetainers contain small amounts of air, which contaminates the gas samples, standards and blanks (Sturm et al., 2015). Because the concentration of CH$_4$ and N$_2$O in air is variable, the contamination also introduced imprecision to sample data. We calculated cut-off concentrations for both gases, below which measured values were considered indistinguishable from those of blank pre-evacuated Exetainers (see Fig. S1 in the Supplement). The determination of the cut-off value took into account both the mean contamination of blank samples and various sources of imprecision in sample data. The cut-off concentrations are approximately equivalent to $9\text{–}19$ and $4\text{–}6$ nM for CH$_4$ and N$_2$O, respectively (exact values are salinity dependent). Precision, which includes errors introduced during transfer of gas samples between syringes and the variation in sample volumes in relation to the residual air in the Exetainers, of all reported data above the cut-off is $<5\%$ relative standard deviation (RSD), as determined by triplicate analysis of all samples. In terms of accuracy, the residual air in the Exetainers caused an average underestimation of 0.9\% in CH$_4$ samples (full range for all reported samples $-2.6$ to $11.4\%$) and a 1.1\% overestimation in N$_2$O samples (full range $-2.7$ to $10.1\%$). Diffusive exchange through the plastic of the syringes was found to be negligible ($<1\%$). The errors inherent to the method do not affect any of the main results or conclusions presented here. For a detailed description of the error and the establishment of the cut-off, we refer the reader to the Supplement.

3 Results

All stations exhibited strong salinity stratification on all sampling occasions, with clear differences between the surface and bottom water salinities (Fig. 2a). The halocline was typically at 60–80 m depth. We observed a net increase in the bottom water salinities at BY15, BY20, BY32 and BY38 from March to December. Absolute bottom water salinities displayed large variation between 308 and 726 nM (Fig. 2c), with the CH$_4$-enriched deep water mass reaching maximum of 230 nM at 225 m depth. In the WGB at BY32 and BY38, no oxygen was detected below the halocline at any time (Fig. 2b), and H$_2$S was present in all bottom water samples. Bottom water salinities were found to be indistinguishable from those of blank pre-evacuated Exetainers (see Fig. S1 in the Supplement). The determination of the cut-off value took into account both the mean contamination of blank samples and various sources of imprecision in sample data. The cut-off concentrations are approximately equivalent to $9\text{–}19$ and $4\text{–}6$ nM for CH$_4$ and N$_2$O, respectively (exact values are salinity dependent). Precision, which includes errors introduced during transfer of gas samples between syringes and the variation in sample volumes in relation to the residual air in the Exetainers, of all reported data above the cut-off is $<5\%$ relative standard deviation (RSD), as determined by triplicate analysis of all samples. In terms of accuracy, the residual air in the Exetainers caused an average underestimation of 0.9\% in CH$_4$ samples (full range for all reported samples $-2.6$ to $11.4\%$) and a 1.1\% overestimation in N$_2$O samples (full range $-2.7$ to $10.1\%$). Diffusive exchange through the plastic of the syringes was found to be negligible ($<1\%$). The errors inherent to the method do not affect any of the main results or conclusions presented here. For a detailed description of the error and the establishment of the cut-off, we refer the reader to the Supplement.

4 Discussion

4.1 Spatial impacts of the MBI

Major Baltic inflows usually progress northwards through the EGB, encircle the island of Gotland counterclockwise, and finally move southwards through the WGB (Meier, 2007; Lessin et al., 2014). However, the MBI of 2014 had not reached the WGB by the end of 2015. Although bottom water salinities increased in the WGB during 2015 (Fig. 2a),
Figure 2. Bilinearly interpolated profiles of (a) CTD salinity, (b) CTD oxygen (µM, shading denotes presence of H₂S), (c) [CH₄] (nM) and (d) [N₂O] (nM) from all sampling stations at all sampling times of the survey. The x axis covers the period 17 March to 13 December 2015. Black notches in (a, b) indicate sampling dates, while the narrower bars marked with an asterisk represent conditions in June 2014, prior to the MBI. Circles in columns (c, d) represent gas sampling depths and times, while infill colour is the non-interpolated original concentration. Note that all CH₄ and N₂O data below the cut-off were set to cut-off prior to interpolation (see Supplement S1 for details).
the increase was continuous throughout the year and not accompanied by changes in oxygen conditions. The changes observed in the salinity of the WGB early in the year were therefore likely unrelated to the inflow. Based on 16 years of monitoring data (shown in Fig. S3 in the Supplement), the WGB experiences annual to multi-annual cycles of shoaling and deepening of the halocline, which are related to downward erosion of the halocline by winter storms (Reissmann et al., 2009). The resulting changes in salinity and oxygen conditions in the 60–100 m depth interval dictate the distribution of CH$_4$ (Jakobs et al., 2014) and N$_2$O in the WGB. [CH$_4$] is higher below the halocline, while [N$_2$O] is higher above it (Fig. 2c, d).

In the EGB, in contrast, large impacts of the 2014 MBI could be observed throughout 2015. Already by March, oxic water and noticeable changes in salinity were detected in the deepest part of the Eastern Gotland Basin (site BY15). These initial signals were a combination of new inflowing water and water pushed out of the Bornholm Basin ahead of the inflow (Schmale et al., 2016), as also occurred during the MBI of 2003 (Feistel et al., 2003). Over the following months, a distinct mass of saline, oxic inflow water accumulated in the EGB. At BY10 and BY15, oxygen was present below the halocline for much of 2015. However, [O$_2$] declined again towards the end of the year (Fig. 2b). Such a decline was also observed following the 2003 MBI (Walter et al., 2006), and indicates that the capacity of MBIs to ventilate the Baltic is short-lived. Introduced oxygen is expected to be consumed simultaneously by a range of electron donors, including hydrogen sulfide (H$_2$S), ammonium, reduced forms of manganese and iron, and CH$_4$, which are all present in the stagnant deep waters. Both the physical effects of the inflow (displacement of water masses) and the subsequent evolution of redox conditions throughout 2015 had strong impacts on the distribution of CH$_4$ and N$_2$O at the EGB sites (BY10, BY15 and BY20).

4.2 CH$_4$ dynamics in the EGB

Upon arrival of an MBI into the EGB, the former bottom water mass is typically displaced vertically upwards and northwards (Reissmann et al., 2009). This displacement of CH$_4$-rich stagnant deep water by CH$_4$-poor inflow water may deplete the inventory of CH$_4$ in the water column of the EGB. However, due to turbulent mixing at the contact between the inflow and older water masses (Schmale et al., 2016), oxidation of CH$_4$ may also be expected to accelerate the depletion of CH$_4$. In 2015, we observed the gradual erosion of a mid-water [CH$_4$] maximum between March and August, followed by a re-accumulation of CH$_4$ in both mid- and deep water towards the end of the year (Fig. 2). To estimate the relative effects of displacement and oxidation on the CH$_4$ inventory during the studied period, we compared the CH$_4$ inventory with that of phosphate (PO$_4^{3-}$) at BY15 (Fig. 3). Changes in deep water [PO$_4^{3-}$] in the EGB following an MBI have been shown to be predominantly controlled by displacement (Schneider, 2011), due to the fact that a stagnant PO$_4^{3-}$-rich bottom water mass is displaced by a PO$_4^{3-}$-poor inflow, and comparatively little PO$_4^{3-}$ is sequestered into sediments over the following months despite the expansion of oxic conditions. The logic of the apparent slow response in the redox-sensitivity of PO$_4^{3-}$ is that PO$_4^{3-}$ sequestration requires the presence of solid-phase iron oxyhydroxides, which are not available in the open water column where the majority of the PO$_4^{3-}$ is located. Accordingly, Schneider (2011) estimated that two-thirds of the decline in deep water [PO$_4^{3-}$] in the EGB following the MBI of 2003 could be attributed to displacement.

In the hypothetical scenario in which the decline in both CH$_4$ and PO$_4^{3-}$ inventories from March to August 2015 was controlled by displacement only, the molar ratio of PO$_4^{3-}$ : CH$_4$ would be expected to remain constant. Instead, we observe that the inventory ratio of PO$_4^{3-}$ : CH$_4$ declined rapidly during this period, implying a far stronger impact of oxidation on CH$_4$ than on PO$_4^{3-}$. Indeed, the CH$_4$ inventory was depleted from ~18 to ~3 mMol m$^{-2}$ over this interval, while the PO$_4^{3-}$ inventory declined only slightly, from ~480 to ~400 mMol m$^{-2}$ (Fig. 3). These results provide strong ev-

![Figure 3](image-url)
idence for significant aerobic oxidation of CH$_4$ in the water column as a consequence of the MBI, which concurs with a recent study in which in situ oxidation rates were measured (Schmale et al., 2016).

When expressed as volume-weighted averages (below 70 m), the observed decline in [CH$_4$] over March to August 2015 was from 108 nM in March to 16.3 nM in August. This is a period of 141 days, giving a total rate of loss of 0.65 nM d$^{-1}$, of which 0.51 nM d$^{-1}$ (79%) is potentially due to oxidation, based on the results shown in Fig. 3. Schmale et al. (2016) reported CH$_4$ oxidation rates of up to 0.9 nM d$^{-1}$ and elevated cell counts for methane-oxidizing bacteria at central EGB in March 2015. Such oxidation rates are 2–10 times higher compared to stagnation conditions (Jakobs et al., 2014). The introduction of a second, deeper redoxcline and active turbulent mixing processes clearly accelerate CH$_4$ oxidation following an MBI by enhancing the volume of water in which CH$_4$ and O$_2$ come into contact. It should be noted, however, that the Schmale et al. (2016) study focused specifically on high turbulence transition zones and thus reports localized maximum estimates of CH$_4$ oxidation rates. In contrast, our study provides a first-order estimate of the bulk oxidation rate the sub-halocline water column at site BY15.

4.3 N$_2$O dynamics in the EGB

Under stagnation conditions, the deep anoxic waters of the EGB are almost entirely devoid of N$_2$O (Bretar and Rheinheimer, 1991), but the hypoxic margins are hotspots for N$_2$O production, similar to oceanic oxygen minimum zones (Babbin et al., 2015; Löscher et al., 2012). The accumulation of N$_2$O over time after an MBI is related to the formation of large hypoxic water masses. N$_2$O production is highest under nearly anoxic conditions, where oxygen concentrations restrain both nitrification (too little oxygen) and denitrification (too much oxygen). Both processes produce N$_2$O, with a higher N$_2$O yield in oxygen stress (Goreau et al., 1980; Patureau et al., 1994; Ji et al., 2015). In addition, when nitrifying microbes become oxygen limited, they too start to reduce NO$_2^-$ to N$_2$O (Poth and Focht, 1985; Wilson et al., 2014).

For the most part, the N$_2$O concentrations measured in the oxic waters of this study were between 10 and 20 nM, which is well in agreement with previous studies (Bange, 2006; Walter et al., 2006). The volume-weighted average [N$_2$O] below 70 m depth at BY15 increased from 11.8 to 24.4 nM from March to August (141 d), giving a net increase rate of 0.09 nM d$^{-1}$. Walter et al. (2006) reported a similar increase rate of 0.105 nM d$^{-1}$ from the whole water column below 70 m in the EGB, over a period of 167 days after the 2003 MBI, which they ascribed largely to nitrification. We observed a decline in [NH$_4^+$] in the mid-water layer at BY15 over the period March to August 2015 (Fig. 4). The decline in [NH$_4^+$] resembled the loss of CH$_4$ over the same period, and was coupled to increasing [NO$_2^-$] and [NO$_3^-$] (Fig. 4).

These observations strongly suggest that oxidation of NH$_4^+$ (i.e. nitrification) following the MBI was the main pathway of N$_2$O production during the first half of 2015.

One of the most interesting observations in this study was the extremely high [N$_2$O] (877 nM) measured at 225 m at BY15 in October, which is to our knowledge the second highest value ever reported from the Baltic Sea (Rönner, 1983, reported 1523 nM at BY38 bottom water in the WGB), and several times higher than the concentrations typically found from the Baltic (Bretar and Rheinheimer, 1991; Bange et al., 1996; Bange, 2006; Walter et al., 2006). The oxygen concentration at the depth of the extreme [N$_2$O] in this study was below 1 µM, which is comparable to values previously observed in settings of high N$_2$O production elsewhere in the ocean (Naqvi et al., 2010; Babbin et al., 2015). The large drop in the [NO$_2^-$] and [NO$_3^-$] below 200 m at BY15 from October onwards (Fig. 4) suggests that the rate of benthic denitrification increased towards the end of 2015. Simultaneously increasing [NH$_4^+$] indicate a slowing down of rates of nitrification and possibly enhanced DNRA (Jäntti and Hietanen, 2012). Hence, the transitional conditions between nitrification and denitrification regimes towards the end of 2015 appear to have favoured an extreme, short-lived accumulation of N$_2$O in the deep waters of the EGB. This may be seen as a delayed, but important consequence of the MBI on nitrogen cycling in the Baltic.

4.4 Processes at the northern limit of the MBI

As MBIs progress northwards, the density differences between the old and new water masses are weakened and the interactions become more complex and difficult to predict (Eilola et al., 2014). Inflowing water masses detach into intrusive layers (Baines, 2001) which may interact chaotically under turbulent flow. Site BY20 is situated near the northern margin of the EGB and represents the northernmost limit of the observable effects of the 2014 MBI. In this zone, various
5 Conclusions

The major Baltic inflow of 2014 caused considerable changes in oxygen conditions of the Eastern Gotland Basin, which had extensive effects on the CH$_4$ and N$_2$O dynamics of the basin. CH$_4$ mostly disappeared from the eastern basin during the first half of 2015, mainly due to oxidation following turbulent mixing between old and new water masses. However, CH$_4$ began to accumulate again by the end of the year, as deep water conditions reverted to anoxia. Enhanced N$_2$O production was evident throughout the Eastern Gotland Basin during 2015, attributed primarily to nitrification as a consequence of the MBI. Extreme values of N$_2$O near the seabed in late 2015 were likely caused by a combination of nitrification and denitrification under transitional conditions. The northern limit of the effect of the MBI on CH$_4$ and N$_2$O dynamics appears to have been the northern part of the Eastern Gotland Basin, and even here direct oxygenation of the deep waters was not observed. The Western Gotland Basin CH$_4$ and N$_2$O biogeochemistry was not influenced by the inflow at any point during the study period.

Data availability. Data are available at https://doi.pangaea.de/10.1594/PANGAEA.880534 (Myllykangas et al., 2017).

The Supplement related to this article is available online at https://doi.org/10.5194/esd-8-817-2017-supplement.

Competing interests. The authors declare that they have no conflict of interest.

References


Patureau, D., Davison, J., Bernet, N., and Moletta, R.: Denitrification under various aeration conditions in Commonas


