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Lattice-matched Cu$_2$ZnSnS$_4$/CeO$_2$ solar cell with open circuit voltage boost

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We report a reproducible enhancement of the open circuit voltage in Cu$_2$ZnSnS$_4$ solar cells by introduction of a very thin CeO$_2$ interlayer between the Cu$_2$ZnSnS$_4$ absorber and the conventional CdS buffer. CeO$_2$, a non-toxic earth-abundant compound, has a nearly optimal band alignment with Cu$_2$ZnSnS$_4$ and the two materials are lattice-matched within 0.4%. This makes it possible to achieve an epitaxial interface when growing CeO$_2$ by chemical bath deposition at temperatures as low as 50°C. The open circuit voltage improvement is then attributed to a decrease in the interface recombination rate through formation of a high-quality heterointerface. Published by AIP Publishing. [http://dx.doi.org/10.1063/1.4971779]

Cu$_2$ZnSn(S,Se)$_4$ compounds are among the materials that are currently considered as potentially suitable for terawatt-scale solar energy production. The pure-sulfide material Cu$_2$ZnSnS$_4$ (CZTS) is particularly interesting because it avoids the relatively scarce element Se, and it has a larger band gap (1.5 eV), which is appropriate both for a single-junction solar cell and for a top absorber in a tandem solar cell architecture. Even though its highest reported power conversion efficiency so far (9.1%) was achieved with a CdS heterointerface partner, or buffer layer, interface recombination is still the dominant voltage loss mechanism in the solar cell, which suggests that CdS is not the ideal partner of CZTS. Despite promising results achieved with (Zn,Cd)S and (Zn,Sn)O buffer layers, no alternative materials to date have outperformed the highest efficiency or the highest open circuit voltage achieved with a CdS heterojunction partner.

The open circuit voltage $V_{oc}$ of a solar cell limited by interface recombination can be expressed as

$$V_{oc} = \frac{E_i}{q} - \frac{kT}{q} \ln \left( \frac{qS_pN_i}{J_{sc}} \right)$$

(1)

for a p-type absorber with an inverted heterointerface, i.e., with holes as minority carriers at the interface. $E_i$ is the interface band gap, $q$ is the elementary charge, $k$ is the Boltzmann’s constant, $T$ is the temperature, $N_i$ is the effective density of states in the valence band, $J_{sc}$ is the short circuit current, and $S_p$ is the interface recombination velocity of holes. By modeling interface recombination as Shockley-Read-Hall recombination through a single interface defect level, $S_p$ can be expressed as

$$S_p = N_i \sigma_p v_{th},$$

(2)

where $N_i$ is the area density of interface defects, $\sigma_p$ is a cross section describing how efficiently the holes are captured by the defect, and $v_{th}$ is the thermal velocity.

Eqs. (1) and (2) show that decreasing the interface defect density $N_i$ can be a way to improve the open circuit voltage in a solar cell limited by interface recombination. By analogy to high-efficiency solar cell technology based on III–V semiconductors, this can be achieved by ensuring epitaxial growth of the buffer material on the absorber material. In the absence of epitaxial growth, a large density of atomic dislocations will exist at the interface, which leads to formation of allowed electronic states within the interface bandgap, thus increasing $N_i$ and enhancing interfacial Shockley-Read-Hall recombination. Epitaxial growth is facilitated by a small lattice mismatch between the two heterojunction materials. While CdS has a reasonably small lattice mismatch with the absorber materials Cu(In,Ga)Se$_2$ (~1.5%) and Cu$_2$ZnSnSe$_4$ (~2.4%), its mismatch with CZTS is much larger (~7%). A high value of $N_i$ can therefore be anticipated at the CZTS/CdS interface.

To address this problem, we carried out an initial investigation of the non-toxic, earth abundant material CeO$_2$ as a heterojunction partner of CZTS. CeO$_2$ has a nearly perfect lattice match with CZTS (lattice constant of CZTS: 5.43 Å; lattice constant of CeO$_2$: 5.41 Å; thus a lattice mismatch under 0.4%). This opens the possibility for a high-quality epitaxial interface, which may alleviate the interface recombination problem. CeO$_2$ also has a wide indirect bandgap of 3.3 eV, which minimizes parasitic light absorption. We note that, based on Eq. (1), the lattice mismatch at the CeO$_2$/CdS interface is not expected to limit $V_{oc}$, since $E_i$ is much larger at the CeO$_2$/CdS interface than at the CZTS/CeO$_2$ interface.

CZTS films were prepared by co-sputtering Cu/ZnSn/Sn precursors on Mo-coated soda lime glass using a magnetron.
Cu/Sn ratio. These precursors were then annealed using Rapid Thermal Processor (AS-One 100) in a S- and Sn-containing atmosphere at 560°C. The S atmosphere compensates for S substoichiometry in the precursors and the addition of Sn stabilizes CZTS against decomposition reactions. CeO2 films were deposited on CZTS by chemical bath deposition (CBD) at 50°C in a weakly acidic solution (pH: 6) containing 10 mM Ce(CH3COO)3 and 5 mM KClO3 in Milli-Q water under mild stirring. The expected chemical reactions resulting in thin film deposition are those presented in previous work. CdS (60 nm) was deposited by CBD with a previously described process. ZnO (60 nm) and indium tin oxide (ITO, 200 nm) layers were deposited by RF magnetron sputtering, followed by an Al contact grid. No anti-reflection coating was applied. The total area of the final cells (0.23 cm²) was defined by mechanical scribing. Three different solar cell architectures were fabricated in this work as shown in Fig. 1. In the first architecture (A1), CdS is completely replaced by a stand-alone 30 nm CeO2 buffer layer deposited with a 100 min process. In the second architecture (A2), a thin CeO2 layer of estimated thickness between 1 and 5 nm is inserted between CZTS and the standard CdS buffer layer using a 10–20 min deposition process. The reference architecture (Fig. 1) is a conventional CZTS solar cell structure without CeO2. Only very weak n-type conductivity has been reported for CeO2 and, in fact, the resistivity of the films deposited from a solution with a KClO3 concentration of 10 mM or above were easily peeled off by scotch tape. However, already at a KClO3 concentration of 5 mM, the films were strongly adherent with no peel-off by repeated scotch tape application. A KClO3 concentration of 5 mM was therefore chosen for the deposition process. The maximum CeO2 thickness on CZTS that could be achieved in a single chemical bath deposition run was about 30 nm with a 100 min process (Fig. S1(a), supplementary material).

Before discussing the solar cell results, we want to answer some basic questions about the quality of deposited CeO2 films. The first question is adhesion. By depositing CeO2 on glass/Mo substrates, it was found that the films deposited from a solution with a KClO3 concentration of 10 mM or above were easily peeled off by scotch tape. However, already at a KClO3 concentration of 5 mM, the films were strongly adherent with no peel-off by repeated scotch tape application. A KClO3 concentration of 5 mM was therefore chosen for the deposition process. The maximum CeO2 thickness on CZTS that could be achieved in a single chemical bath deposition run was about 30 nm with a 100 min process (Fig. S1(a), supplementary material).

The second question is whether the deposited film indeed consists of the desired CeO2 material. The fast Fourier transform (FFT) of cross-sectional TEM images of the deposited film yields a pattern that is compatible with CeO2 and not with the main competing phase Ce2O3 (Fig. 2(c)). However, the TEM analysis involves very small regions (few nm). Raman spectroscopy was then performed over a much larger analysis area (about 2 μm diameter). Only one additional Raman peak at 461 cm⁻¹ was revealed on CZTS/CeO2 bilayers compared to a bare CZTS spectrum (Fig. S2, supplementary material). The peak corresponds to the first-order-allowed Raman mode of CeO2, with a small red shift due to size effects. Finally, XPS characterization was performed over an even larger analysis area (about 400 μm diameter). All the XPS peaks corresponding to Ce 3d core levels were fitted and attributed to either CeO2 or Ce2O3 according to reference spectra (Fig. S3, supplementary material). The fraction of CeO2 present in the deposited film is estimated as 70.4% with this method. Hence, some Ce2O3 inclusion should be expected. Since Ce2O3 is not lattice-matched to CZTS, and it has a band gap of almost 1 eV lower than that of CeO2, we assume that Ce2O3 inclusions promote interface recombination and reduce the open circuit voltage enhancement that could be achieved with a pure CeO2 layer.

The third question is whether the deposited film provides a complete coverage of the underlying CZTS layer. SEM images of a 30 nm-thick CeO2 film (Fig. 3) show some non-uniform coverage in correspondence of CZTS grain boundaries and some smaller isolated dips in the CeO2 film profile, which may be interpreted as pinholes. The area...
The fraction of CZTS that is covered by CeO$_2$ can be estimated by comparing the peak intensity of Ce, Cu, Zn, and Sn core levels by XPS, as explained in the supplementary material. The covered area is then estimated as 62%, 94%, and 95% for the films deposited for 20, 60, and 90 min, with estimated thicknesses of 1–5, 15, and 25 nm, respectively. Thus, it seems as if the very thin films (20 min deposition time) employed in architecture A2 may have rather poor coverage. However, in that case the CeO$_2$ thickness is comparable to the XPS probing depth (about 2 nm). Therefore, part of the CZTS signal is likely to originate from CZTS buried under CeO$_2$ so that the covered fraction is in reality higher. The consequence of incomplete coverage is simply the coexistence of CZTS/CeO$_2$ and CZTS/CdS heterojunctions in parallel to the solar cells.

The fourth question is whether CeO$_2$ forms an epitaxial interface with CZTS. It is well known that a moderate spike-like conduction band offset (CBO) at the heterointerface, in the 0–0.4 eV range, is optimal as it reduces interface recombination without blocking photocurrent transport. The valence band offset (VBO) was estimated by XPS as shown in Fig. 4. The experimental (bulk) band gaps of the two materials were then added to the VBO to finally obtain a CBO of $-0.12 \pm 0.20$ eV. For the CeO$_2$ indirect gap, we used a value of $3.3 \pm 0.1$ eV measured by the internal photoemission yield. The extracted CBO is only slightly below the optimal range, and it is actually more favorable than the previously measured CZTS/CdS band offset. Considering the many possible sources of error in the measurement, including the use of as-deposited surfaces for analysis, this is considered as a promising result. We also note that the CeO$_2$ Fermi level lies about 2.7 eV above the valence band, which indicates that CeO$_2$ is n-type in the analysis region, similar to previous reports.

![FIG. 2. (a) Cross-sectional HRTEM image across the interface between CeO$_2$ (top) and CZTS (bottom). (b) FFT of a selected region across the interface. A reflection at 3.13 Å is visible, corresponding to the distance between (111) planes of CeO$_2$ and between the (112) planes of CZTS. (c) FFT of a selected region within CeO$_2$. The diffractogram can be indexed as a CeO$_2$ crystal along with [103] as the zone axis. (d) FFT of a selected region within CZTS. Reflections corresponding to the (112) and (310) planes are visible.](image)

![FIG. 3. The cross sectional SEM image of a CeO$_2$ film grown on CZTS used for architecture A1. The film is about 30 nm thick, which is the maximum thickness that could be achieved with a single CBD process in this work. Some scattered pinholes and inhomogeneous coverage of the grain boundaries are seen.](image)

![FIG. 4. CZTS/CeO$_2$ band alignment extracted by XPS. The valence band maximum (VBM) of CZTS with respect to the Fermi level is $-0.60 \pm 0.10$ eV in a bare CZTS sample and $-0.75 \pm 0.10$ eV in a CZTS sample coated with a thin CeO$_2$ layer (20 min deposition time, red circles). From this, band bending (BB) in CZTS is estimated as $0.15 \pm 0.10$ eV. This is also confirmed from the average shift in the Cu 2p, Zn 2p, and Sn 3d core levels between the bare CZTS sample and the CeO$_2$/CZTS sample. Shifting the XPS spectrum of the bare CZTS sample by the BB (black line) allows deconvolution of the CeO$_2$ valence band signal (blue triangles), located $-2.67 \pm 0.10$ eV below the Fermi level. The shifted spectrum of bare CZTS (black line) fits well with the CeO$_2$/CZTS spectrum until the onset of the CeO$_2$ valence band, as expected.](image)
The current-voltage characteristics of the three best solar cells fabricated with the three different architectures are shown in Fig. 5. A small improvement in open circuit voltage is achieved by architecture A2 with respect to the reference architecture. The accompanying degradation in short circuit current will be discussed in the following. Interestingly, the statistics shown in Fig. 6 indicate that the open circuit voltage boost is reproducible over two separate batches (i) and (ii) regardless of precursor composition (C1 or C2). In all those cases, the open circuit voltage of the solar cells with A2 architecture is always between 20 mV and 100 mV higher than in the corresponding solar cells with the standard CdS architecture. Additional statistics in Fig. S6 (supplementary material) confirm this trend. The highest open circuit voltage achieved in this study with the A2 architecture was 641 mV, and the highest efficiency was 6.6% (Fig. 5). Conversely, complete current blocking and no photovoltaic effect was observed in the solar cells with a stand-alone CeO2 buffer layer (architecture A1 in Fig. 5).

To interpret these results, we refer to the band structure of CeO2. What has been referred to as the “conduction band” in this work is a band consisting of highly localized Ce 4f states, 3.3 eV above the valence band as mentioned before. However, localization of states in this band implies that the electron effective mass is very high, and hence the electron mobility is very low. This can be inferred by a nearly complete lack of dispersion of those states in reciprocal space. For this reason, a band consisting mostly of Ce 5d states, which lies about 6 eV above the valence band has lighter effective masses, is often quoted as the conduction band of CeO2. From our band alignment study (Fig. 4), we found that the 4f band has a nearly optimal CBO with the CZTS conduction band, while the 5d band lies at a much higher energy. Then, we can attribute the complete lack of photocurrent in architecture A1 to the very poor transport properties of the 4f band. A 30 nm CeO2 layer is therefore thick enough to completely impede electron transport. Instead, when CeO2 is very thin as in architecture A2 (1–5 nm), short circuit current and fill factor losses are greatly diminished and in some cases eliminated (Fig. 5 and S5, supplementary material). This is possibly due to tunneling-based transport between the CZTS and CdS conduction bands through the thin interlayer. Residual current losses, as in Fig. 5, are probably due to thickness inhomogeneity of CeO2, or to the fact that an even thinner layer is necessary. Growth of CeO2 by atomic layer deposition may be beneficial in this respect.

We conclude that inserting a thin lattice-matched CeO2 layer between CZTS and CdS alleviates interface recombination and results in a reproducible open circuit voltage boost in the solar cell. This is attributed to the formation of a high-quality CZTS/CeO2 heterointerface, with instances of epitaxial growth observed in some regions. We suggest that the open circuit voltage could be further improved if epitaxy could be obtained on a larger scale, if CeO2 inclusions could be decreased, and if the surface coverage could be improved. Etching the CZTS surface immediately prior to CeO2 deposition may facilitate the formation of an epitaxial interface. It should be kept in mind that the very large electron effective mass in the CeO2 conduction band puts a severe constraint on the maximum thickness of the CeO2 film, which should only be a couple of nm thick in order to avoid the dramatic current losses.

See supplementary material for additional TEM images, phase analysis of the CeO2 layer, compositional analysis across the interface, estimation of CeO2 coverage, and more detailed statistics on the solar cell parameters for the reference architecture and architecture A2.

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